

## RARE METALS AND POLITICS

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**Abstract:** Rare earth elements are a collection of 17 elements, which include lanthanides and scandals. They are used to make many devices – from electronics to the defense industry. Currently, China is the largest producer of these raw materials, although deposits of these elements are also found in other countries, e.g. United States of America or Japan. Although the name suggests otherwise, metal deposits are abundant in the world. The problem is their extraction and concentration of individual minerals, resulting from the chemical structure.

**Key words:** rare earth elements, rare earth metals, ores, resources, demand

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### INTRODUCTION

Rare earth metals, also known as REE (rare earth elements), are a group of 17 elements, which include lanthanides and scandals. This group includes elements with atomic numbers from 57 to 71 (15 lanthanides: La – Lanthanum, Ce – Cerium, Pr – Praseodymium, Nd – Neodymium, Pm – Promethium, Sm – Samarium, Eu – Europium, Gd – Gadolinium, Tb – Terbium, Dy – Dysprosium, Ho – Holmium, Er – Erbium, Tm – Thulium, Yb – Ytterbium, Lu – Lutetium) and 21 and 39 atomic scandals (Sc – Scandium, Y – Yttrium), which have similar chemical properties (figure 1) (Hedrick, 1997).

These elements belong to the group of rock-loving elements – they concentrate mainly in the Earth's crust. They are also incompatible, which means that during the transformation of rocks or crystallization of magma they accumulate in the liquid phase – the alloy. This feature decreases with the ionic radius from cerium to lutetium.

REE are characterized by high chemical activity. This results in occurrence in natural conditions only in the form of salts – carbonates, phosphates, silicates or oxides, occurring in associations with other metals (Castor and Hedrick, 2006).

Contrary to popular judgment and name, these elements do not occur in nature rare. The problem is their distribution and concentration. This is due to their atomic structure and geochemical properties. Elements with even atomic

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numbers are more common on Earth than those with odd atomic numbers. Similarly, lighter rare earth elements (LREE) have a larger ion radius, which translates into less frequent substitution with other elements. They also show a greater tendency to occur in mineral phases than the heavier rare earth elements – HREE (occurring from Gd to Lu) (table 1).

**Periodic Table of the Elements**

1 H Hydrogen 1.008																	2 He Helium 4.003
3 Li Lithium 6.941	4 Be Beryllium 9.012											5 B Boron 10.811	6 C Carbon 12.011	7 N Nitrogen 14.007	8 O Oxygen 15.999	9 F Fluorine 18.998	10 Ne Neon 20.180
11 Na Sodium 22.990	12 Mg Magnesium 24.305											13 Al Aluminum 26.982	14 Si Silicon 28.086	15 P Phosphorus 30.974	16 S Sulfur 32.066	17 Cl Chlorine 35.453	18 Ar Argon 39.948
19 K Potassium 39.098	20 Ca Calcium 40.078	21 Sc Scandium 44.956	22 Ti Titanium 47.88	23 V Vanadium 50.942	24 Cr Chromium 51.996	25 Mn Manganese 54.938	26 Fe Iron 55.933	27 Co Cobalt 58.933	28 Ni Nickel 58.693	29 Cu Copper 63.546	30 Zn Zinc 65.39	31 Ga Gallium 69.723	32 Ge Germanium 72.61	33 As Arsenic 74.922	34 Se Selenium 78.09	35 Br Bromine 79.904	36 Kr Krypton 84.80
37 Rb Rubidium 84.468	38 Sr Strontium 87.62	39 Y Yttrium 88.906	40 Zr Zirconium 91.224	41 Nb Niobium 92.906	42 Mo Molybdenum 95.94	43 Tc Technetium 98.907	44 Ru Ruthenium 101.07	45 Rh Rhodium 102.906	46 Pd Palladium 106.42	47 Ag Silver 107.868	48 Cd Cadmium 112.411	49 In Indium 114.818	50 Sn Tin 118.71	51 Sb Antimony 121.760	52 Te Tellurium 127.6	53 I Iodine 126.904	54 Xe Xenon 131.29
55 Cs Cesium 132.905	56 Ba Barium 137.327	57-71 Lanthanides	72 Hf Hafnium 178.49	73 Ta Tantalum 180.948	74 W Tungsten 183.85	75 Re Rhenium 186.207	76 Os Osmium 190.23	77 Ir Iridium 192.22	78 Pt Platinum 195.08	79 Au Gold 196.967	80 Hg Mercury 200.59	81 Tl Thallium 204.383	82 Pb Lead 207.2	83 Bi Bismuth 208.980	84 Po Polonium [209]	85 At Astatine [209]	86 Rn Radon 222.018
87 Fr Francium 223.020	88 Ra Radium 226.025	89-103 Actinides	104 Rf Rutherfordium [261]	105 Db Dubnium [262]	106 Sg Seaborgium [266]	107 Bh Bohrium [264]	108 Hs Hassium [269]	109 Mt Meitnerium [268]	110 Ds Darmstadtium [269]	111 Rg Roentgenium [272]	112 Cn Copernicium [277]	113 Uut Ununtrium unknown	114 Fl Flerovium [289]	115 Uup Ununpentium unknown	116 Lv Livermorium [293]	117 Uus Ununseptium unknown	118 Uuo Ununoctium unknown
57 La Lanthanum 138.906	58 Ce Cerium 140.115	59 Pr Praseodymium 140.908	60 Nd Neodymium 144.24	61 Pm Promethium 144.913	62 Sm Samarium 150.36	63 Eu Europium 151.964	64 Gd Gadolinium 157.25	65 Tb Terbium 158.925	66 Dy Dysprosium 162.50	67 Ho Holmium 164.930	68 Er Erbium 167.26	69 Tm Thulium 168.934	70 Yb Ytterbium 173.04	71 Lu Lutetium 174.967			
89 Ac Actinium 227.028	90 Th Thorium 232.038	91 Pa Protactinium 231.036	92 U Uranium 238.029	93 Np Neptunium 237.048	94 Pu Plutonium 244.064	95 Am Americium 243.061	96 Cm Curium 247.070	97 Bk Berkelium 247.070	98 Cf Californium 251.080	99 Es Einsteinium [254]	100 Fm Fermium 257.095	101 Md Mendelevium 258.1	102 No Nobelium 259.101	103 Lr Lawrencium [262]			

Figure 1. Periodic Table of elements <sup>1</sup>

Table 1. Light and Heavy rare earth elements  
(Source: UNCTAD Secretariat from the British Geological Survey)

Atomic number	Name	Symbol	LREE / HREE
57	Lanthanum	La	LREE
58	Cerium	Ce	LREE
59	Praseodymium	Pr	LREE
60	Neodymium	Nd	LREE
61	Promethium	Pm	LREE
62	Samarium	Sm	LREE
63	Europium	Eu	LREE
64	Gadolinium	Gd	LREE
65	Terbium	Tb	HREE
66	Dysprosium	Dy	HREE
67	Holmium	Ho	HREE
68	Erbium	Er	HREE
69	Thulium	Tm	HREE
70	Ytterbium	Yb	HREE
71	Lutetium	Lu	HREE
39	Yttrium	Y	
21	Scandium	Sc	

<sup>1</sup> <https://sciencenotes.org/printable-periodic-table-chart/>

The characteristic features of rare earth metals include: high dispersion, no occurrence in pure metallic form, the rarity of creating own minerals visible to the human eye (Porowski and Kaczor–Kurzawa, 2016).

### PRESENCE IN THE WORLD

There are more than 250 REE minerals known in the world, but only some of them are important for industry. In table 2 the most important of them are presented. They are most often found in the lithosphere in the form of fluorocarbonates, phosphates and silicates.

**Table 2.** Main minerals containing REE

Source: Ln – lanthanides group, REO – rare earth oxides (Zhang et al., 2016)

Mineral	Chemical formula	Content of REO (%)
Bastnaesite	$\text{LnCO}_3\text{F}$	74,8
Monazite	$(\text{Ln,Th})\text{PO}_4$	65,1
Xenotime	$\text{YPO}_4$	62,0
Fluocerite	$(\text{Ce,Ln})\text{F}_3$	83,4
Parisite	$\text{CaLn}_2(\text{CO}_3)_3\text{F}_2$	60,3
Fergusonite	$\text{YNbO}_4$	39,9
Gadolinite	$\text{Y}_2\text{FeBe}_2\text{Si}_2\text{O}_{10}$	48,3
Aeschynite	$(\text{Y,Ca,Fe})(\text{Ti,Nb})_2(\text{O,OH})_6$	24,6
Euxenite	$(\text{Y,Ca,Ce})(\text{Nb,Ta,Ti})_2\text{O}_6$	24,3
Synchysite	$\text{Ca}(\text{Y,Ce})(\text{CO}_3)_2\text{F}$	49,6
Samarskite	$(\text{Y,Fe,U})(\text{Nb,Ta})_5\text{O}_4$	24,3
Polycrase	$(\text{Y,Ca,Ce,U,Th})(\text{Ti,Nb,Ta})_2\text{O}_6$	19,5
Loparite	$(\text{Ce,NaCa})_2(\text{Ti,Nb})_2\text{O}_6$	29,8

The content of rare earths in the lithosphere is very diverse and depends mainly on the type of rock and its origin. The content of REE is generally in the range of 0,1 to 100 mg / kg. For example, in rhyolites and granites, there is a higher concentration of these metals than in basalt. Likewise, clays and shales contain more REE than limestone and sandstone. In addition, LREE content in rock ores is generally greater than HREE. In table 3 the number of rare earths is given (Porowski and Kaczor–Kurzawa, 2016).

**Table 3.** Abundance of REE in the Earth's crust in parts per million

(Source: Moran-Palacios et al., 2019)

Element	Kleber and Love (1963)	Ryan (1968)	Jackson and Christiansen (1993)	Wedepohl (1995)	Sabot and Maestro (1995)	McGill (1997)	Lide (1997)	Earth Crust (ppm)
Sc	10	-	-	16	10	5-10	22	
Y	28	-	29	24	28	28-70	33	
La	18	18,3	29	30	18	5-18	39	30
Ce	46	46,1	70	60	46	20-46	66,5	60
Pr	5,5	5,53	9	6,7	5,5	3,5-5,5	9,2	6,7
Nd	24	23,9	37	27	24	12-24	41,5	27
Sm	6,5	6,47	8	5,3	6,5	4,5-6,4	7,05	5,3
Eu	0,5	1,06	1,3	1,3	0,5	0,14-1,1	2	1,3
Gd	6,4	6,36	8	4	6,4	14,5-6,4	6	4
Tb	0,9	0,91	2,5	0,7	0,9	0,7-1	1,2	0,7
Dy	5	4,47	5	3,8	5	4,5-7,5	5,2	3,8
Ho	1,2	1,15	1,7	0,8	1,2	0,7-1,2	1,3	0,8
Er	4	2,47	3,3	2,1	4	2,5-6,5	3,5	2,1
Tm	0,4	0,2	0,27	0,3	0,4	0,1-1	0,52	0,3
Yb	2,7	2,66	0,33	2	2,7	2,7-8	3,2	2
Lu	0,8	0,75	0,8	0,7	0,8	0,8-1,7	0,8	0,7

About 67% of the world's resources occur in three regions: China, which has 31% of the resources of various types – bastnaesite (Sichuan and Inner Mongolia), laterite clays (Jiangxi) and monacite black beach sands. In Russia, there is 21% of the world's stocks of shovel, while in the USA – 15%, mainly bastnaesite and monacite. Australia also has small ore deposits – 6% (monocytite deposits), and Canada and India 1% each (Smakowski et al., 2015).

Other sources of rare earth metals are xenotime ores, found in Malaysia and Thailand, phosphorites, apatites, eudialite and waste of uranium solutions, located in Kazakhstan and Russia (Smakowski et al., 2015).

Data on the volume of global rare earths resources vary quite significantly. According to Smakowski et al., they amount to 88 million tons of  $\text{Ln}_2\text{O}_3$ . In turn, Cahus-Moszkowski and Białecka (2012) estimate them at about 110 million tons.

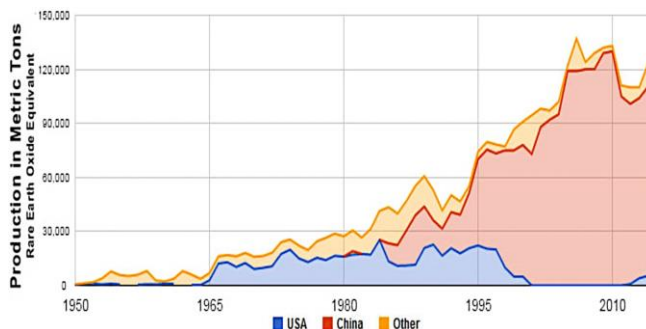
## PRODUCTION

Rare earths are usually obtained from three types of ores: bastnaesite (USA, China), monacite (China, USA, Australia, Brazil, India, Malaysia, Sri Lanka and Thailand) and laterite. The first stage in the production of rare earths is the extraction of ore using standard mining methods. The ore was mined for gravel. It continues to crumble until small fractions of rock are obtained. Further, the crushed ore is placed in a tank, where individual metals attach to the bubbles of admitted air in the filtration process. Rare earths are obtained by using various chemicals. The process is used until pure elements are obtained in the form of oxides (Hurst, 2010).

The demand for rare earths is constantly increasing. In 1950 it was about 100 tons / year. In 1990 it increased to 40,000 tons per year. In 2000 it amounted to around 80,000 tonnes and now - over 200,000 tonnes (Wyłuda, 2016).<sup>2</sup>

Prior to 1965, the demand for rare earths was relatively low. At that time, most of the world's supplies were produced from crumb deposits in India and Brazil. In the 1950s, South Africa became a leading producer of monazite deposits containing rare earths. At that time, the Mountain Pass Mine in California produced small amounts of rare earth oxides from Precambrian carbonate.<sup>3</sup>

The demand for rare earths first exploded in the mid-1960s, when the first color televisions went on sale. Europium was a necessary raw material to produce color images (figure 2).



**Figure 2.** Production of rare earth elements  
(Source: chemical-materials.elsevier.com)

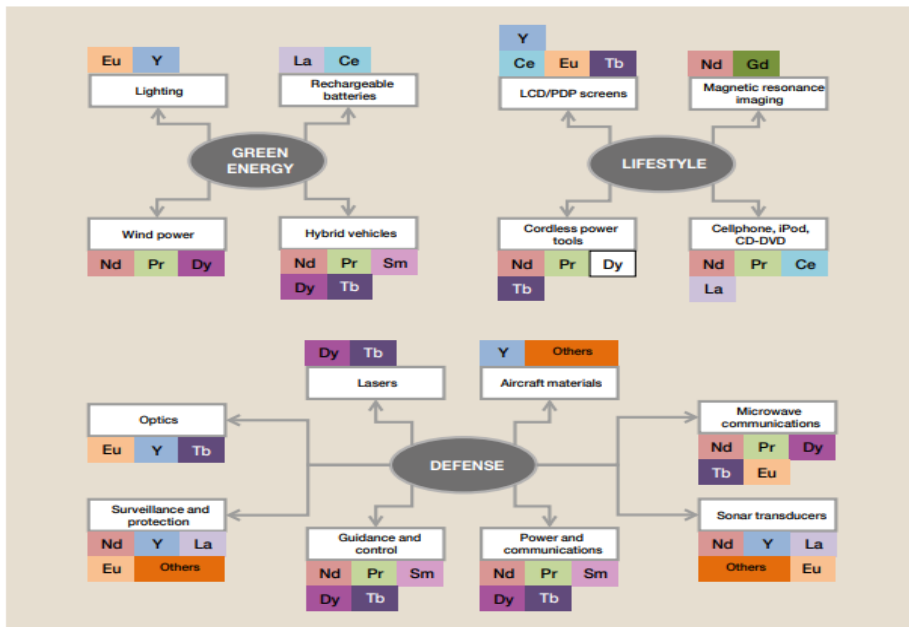
<sup>2</sup> <https://www.edukacjagieldowa.pl/2016/09/cenniejsze-niz-zloto-diamenty-metale-ziem-rzadkich/>

<sup>3</sup> <https://poltimes.pl/metale-ziem-rzadkich-strategiczne-surowce-w-rekach-chin/>

At the Mountain Pass Mine, production of europium from bastnaesite began, which contained about 0,1% of europium. As a result, the Mountain Pass Mine has become the largest producer of rare earths in the world, and the United States has become a leading producer (Lifton, 2010).

In the early 1980s, China began producing significant amounts of rare earth oxides. They became a world leader in the early 1990s. In the 1990s and early 2000, China successively strengthened its importance in the world economy of rare earth oxides. Prices of Chinese rare earths were so low that Mountain Pass Mine and many others around the world were unable to compete and stopped functioning (Hurst, 2010).

At the same time, global demand has risen sharply because rare earths have been used in the production of safety, aviation and automotive components, industry and consumer electronics (figure 3).



**Figure 3.** Sectors using REE

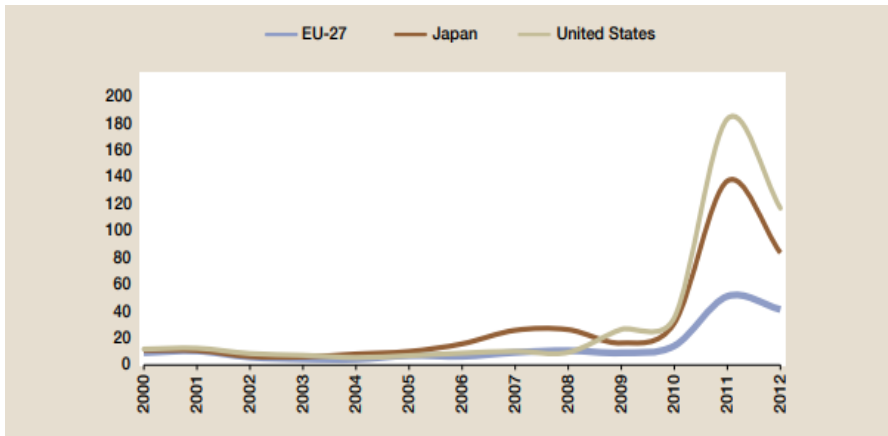
(Source: UNCTAD Secretariat from Great Western Minerals Group Ltd)

China has used its dominance and began to restrict exports and made the rare earth oxides prices reached the highest level ever (figure 4) (Morrison et al., 2012).<sup>4</sup> China's Rare Earth Industry and Export Regime: Economic and Trade Implications for the United States. Congressional Research Service).

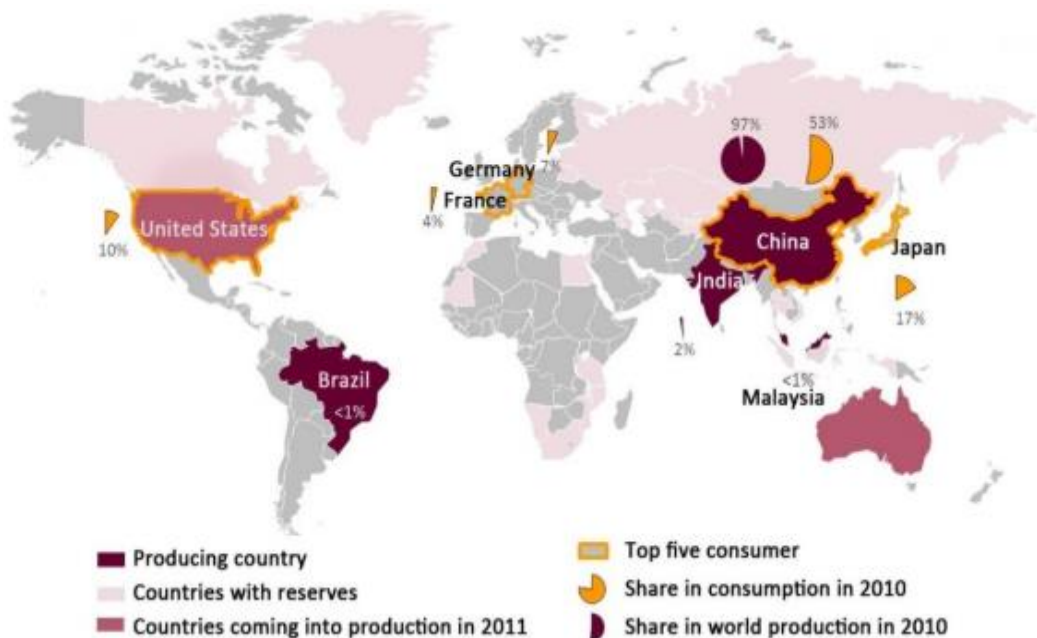
Apart from being the world's largest producer of rare earths, China is also the largest consumer (figure 5). China uses REE primarily to produce electronic products for domestic and foreign markets. Japan and the United States are the next largest consumers of the elements. To protect the added value of their manufacturing sector, China is reluctant to sell rare earths.<sup>5</sup>

<sup>4</sup> <http://www.fas.org/sgp/crs/row/R42510.pdf>

<sup>5</sup> <https://independenttrader.pl/metale-ziem-rzadkich-marginalizowane-aktywo.html>



**Figure 4.** Changes of unit value of rare earth imports in 2000-2012 (US\$ per kilo)  
(Source: UNCTAD Secretariat from UN COMTRADE (HS 2012: 280530))



**Figure 5.** Map of global producers and consumers of REE <sup>6</sup>

The climax of Chinese domination could have occurred in 2010, when China oversaw around 95% of world production of rare earth metals. This was a stimulus for mining companies in the United States, Australia, Canada and other countries that began to reassess old prospects for rare earths and seek new ones. High prices led manufacturers to take three steps (Pourmand et al., 2012):

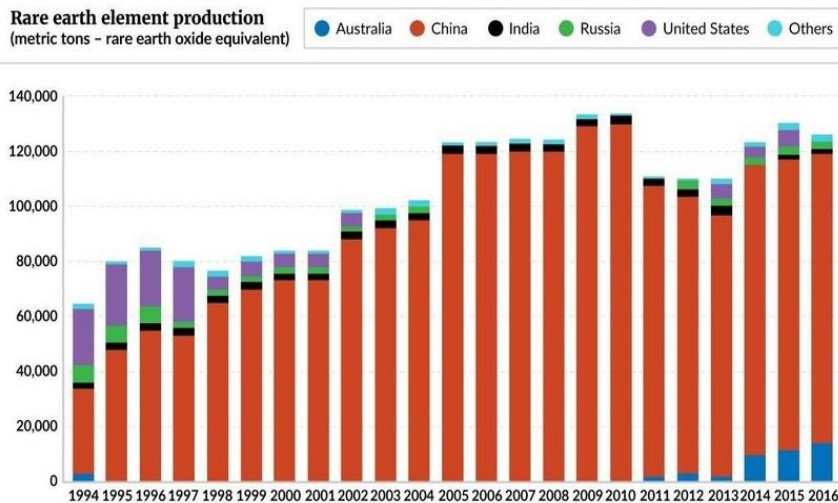
- looking for ways to reduce the amount of rare earths used to make each of their products;
- looking for alternative materials that replace rare earths;
- develop variant products that do not need rare earths.

<sup>6</sup> <https://ecowasterecycling.wordpress.com/>

Chinese companies buy rare earths in other countries. In 2009, the China Non-Ferrous Metal Mining Company acquired the majority of shares in the Australian company Lynas Corporation, which has one of the highest rare earths production outside China. They also bought the Baluba Mine in Zambia (Kelly, 2009).

In 2011, Australian Mines began producing rare earth oxides. In 2012 and 2013, they were covering around 2–3% of global demand. In 2012, the Mountain Pass Mine returned to production, and in the United States in 2013, about 4% of the world's rare earths was produced. Manufacturing in Brazil, Malaysia, Russia, Thailand and Vietnam remained stable or increased (Zamecki, 2011).

Recently, the United States Geological Survey has identified significant resources outside China. Although China is dominant in the production of rare earth metals, they merely superintend about 36% of global resources. This gives other countries the chance to become important manufacturers, when China doesn't want to sell rare earth metals below production costs. <sup>7 8</sup>



**Figure 6.** Production of REE by country in 1994-2016  
(Source: European Commission/Transport&Environment 2017) <sup>9</sup>

The figure 6 shows the dominance of China in the production of rare earths in the years 1994-2016. The United States was an important manufacturer until the 1990s, but cheap materials being sold by China forced the withdrawal of mines in the United States and other countries. Because China has reduced exports and prices have risen sharply in 2009 and 2010, Australian and US mines have started operating again. The Chinese government explained that this was done to guarantee the supply of rare earths for domestic production and for environmental reasons. This change generated by China has caused panic, and some rare earth prices have risen significantly. Furthermore, Japan, the United States and the European Union filed to the World Trade Organization about China's restrictive trade policies of rare earth elements (Kalantzakos, 2017).

<sup>7</sup> <http://www.marketwatch.com/story/molycorp-strikes-rare-earth-elements-in-california-2011-10-04>

<sup>8</sup> <http://online.wsj.com/article/SB10001424052970204612504576609413994133684.html>

<sup>9</sup> [www.GISreportsonline.com](http://www.GISreportsonline.com)

## CURRENT SITUATION

The demand for rare earths is higher than supply. Due to the ever-increasing demand, it is not known whether producers will be able to increase production before stocks run out. According to OECD forecasts, the demand for rare earths in 2060 is expected to increase to 19 Gt/year.<sup>10</sup>

By introducing export restrictions (including hindering the export of elements from the country), China has strengthened its position in the world. The government explains these restrictions by increasing internal consumption and protecting the environment. Through previous state policy - a prosperous mining industry, low labor costs, a lack of environmental standards and unlicensed mining and processing, China has led to a monopoly on mining.

Although Japan has discovered large deposits of rare earths, it is not known whether the balance of power will change. According to the forecasts, 1 km<sup>2</sup> metals from Japanese deposits could satisfy 20% of the current demand of the raw material. However, mining is a problem. The deposits are located at the bottom of the Pacific Ocean at a depth of 3500 to 6000 m. Until the Japanese start exploiting the deposits, China will continue to use the current situation to exert pressure and negotiate commercial transactions (Błoński, 2011).

Currently in the United States, in California, it operates one ore mine rare earth metals. Mined rocks, however, are sent for processing to China, which is subject to a 25% duty. As a result, China covers 80% of the US demand for rare earths. Suspending supplies would pose a real threat to the US economy. Without them, it is impossible to produce computers, military equipment or electric / hybrid cars. The real threat of the trade war on the Washington-Beijing line was exacerbated last year by cutting off the Chinese group Huawei from American technology.<sup>11 12 13</sup>

## CONCLUSION

Currently, rare earth metals are gaining significant significance for economic development, in addition to oil and natural gas. Without their use, many inventions of modern technologies would look completely different, and others would not exist at all. Rare earth metals are a key component of advanced military and civilian devices, as well as a key element in the use of green technologies such as wind turbines and hybrid cars. 97 percent of rare earths are currently produced in China. However, the growing global demand as well as Chinese export restrictions raise concerns of international corporations as to maintaining supply continuity. Situations are additionally hindered by China's increasing domestic demand.

Global governments, scientists and industry looking for new solutions-extraction, recycling or substitution, but for now it is not enough. Newly discovered deposits are not able to cover the growing demand of the world. For now, China's conditions should be accepted, but alternatives should be sought out.

## Aknoledgments

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<sup>10</sup> <https://www.spidersweb.pl/2019/07/zapotrzebowanie-na-metale-ziem-rzadkich-elektronika-gornictwo-kosmiczne.html>

<sup>11</sup> <https://klubjagiellonski.pl/2019/08/14/metale-ziem-rzadkich-oczko-w-glowie-chinskiej-republiki-ludowej/#>

<sup>12</sup> <https://forsal.pl/artykuly/1414979,metale-ziem-rzadkich-nowym-orezem-w-walce-chin-z-usa.html>

<sup>13</sup> <https://www.wnp.pl/tech/chca-zaspokoic-zapotrzebowanie-na-metale-ziem-rzadkich,346568.html>



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